C1—N7—C8	126.6 (2)	127.40 (13)	-
N7-C8-N10	113.4 (2)	111.28 (13)	116.0 (4)
C8—N10 ⁻ —N7	120.7 (2)	115.15 (13)	115.1 (4)
(<i>b</i>)			
	I-Carbamoyl-	Bis(phenyl)-1-	1-Carbamoyl-
	guanidine	carbamoyl-	guanidinium
	(neutral)	guanidinium	(cation)
		(cation)	
C13—N19	-	1.409 (2)	-
N19-C20	1.358 (4)	1.344 (2)	1.326(3)
C20-021	1.251 (4)	1.216 (2)	1.224 (3)
C20-N22	1.355 (4)	1.402 (2)	1.392 (3)
N22—C23	1.348 (4)	1.349 (2)	1.358 (3)
C23-N24	1.324 (4)	1.311 (2)	1.314 (3)
C23-N25	1.338 (4)	1.325 (2)	1.319 (3)
N25-C26	-	1.432 (2)	-
O21· · · H24A	1.98 (4)	2.02 (2)	2.04 (3)
C13-N19-C20	_	128.72 (14)	-
N19-C20-N22	113.4 (3)	110.64 (13)	113.6 (2)
C20-N22-C23	119.8 (3)	126.91 (13)	125.9 (2)
N22-C23-N25	116.2 (3)	116.64 (13)	117.1 (2)
C23-N25-C26	_	125.78(13)	_
O21-C20-N22	127.1 (3)	122.76 (14)	122.4 (2)
N22—C23—N24	126.3 (3)	120.96 (14)	121.0 (2)
	(-)	• •	• •

Table 2. Hydrogen-bonding geometry (Å, °)

Mean d(N-H) = 0.88 Å; the e.s.d. of an individual measurement is 0.02 Å, and this is also the e.s.d. of the sample of six values. Note that the e.s.d.'s given in Table 3 of YHCT96 did not properly take into account the hydrogen e.s.d.'s.

N—H···O/N—C	$d(N \cdots O/N)$	$d(H \cdot \cdot \cdot ON)$	Angle †	Angle ‡
Intramolecular N24—H24A···O21=C20§	2.712 (2)	2.02 (2)	134 (2)	98.3 (6)
Cation-to-anion	1 999 (1)	2.01.(2)	166 (2)	1247(5)
$N19 - H19 \cdot \cdot \cdot N12 = C11$ $N22 - H22 \cdot \cdot \cdot N12 = C11$	2.943 (2)	2.01 (2)	157 (2)	103.6 (5)
N24—H24 B ···N12'=C11' N24–H24 A ···O21 ⁱⁱ =C20 ⁱⁱ 8	2.979 (2) 3.037 (2)	2.20 (2) 2.38 (2)	147 (2) 134 (2)	114.6 (5) 152.2 (5)
N25H25O9C8	2.813 (2)	1.96 (2)	163 (2)	144.9 (5)
Anion-to-anion N7—H7···N10 ⁱⁱⁱ —C8 ⁱⁱⁱ	3.063 (2)	2.26 (2)	177 (2)	134.9 (5)
\uparrow N—H···O/N. \ddagger H· formed by N24—H24A.	···O/N—C.	§ Bifurcat	ed hydrog	en bond
Symmetry codes: (i) $1 + x$, y, z; (ii) 1 -	-x, -y, -z;	iii) $-x, 1 -$	-y, 1-z.

Program(s) used to solve structure: *SHELXS*86 (Sheldrick, 1990). Program(s) used to refine structure: *SHELXL*93 (Sheldrick, 1993). Molecular graphics: *ORTEPII* (Johnson, 1976).

The authors are grateful to Professor H. J. E. Loewenthal and Dr Mark Botoshansky for their help.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: FG1255). Services for accessing these data are described at the back of the journal.

References

- Begley, M. J., Hubberstey, P., Martindale, S. P., Moore, C. H. M. & Price, N. S. (1988). J. Chem. Res. (S), pp. 2-3.
- Begley, M. J., Hubberstey, P. & Moore, C. H. M. (1985). J. Chem. Res. (S), pp. 380-381.
- Begley, M. J., Hubberstey, P. & Moore, C. H. M. (1986). J. Chem. Res. (S), pp. 120-121.

© 1997 International Union of Crystallography Printed in Great Britain – all rights reserved

- Crystal Structures. University of Göttingen, Germany. Stainton, N. M., Harris, K. D. M. & Howie, R. A. (1991). J. Chem.
- Soc. Chem. Commun. pp. 1781–1784.

Yang, Q.-C., Huang, D.-M., Chen, H.-Y. & Tang, Y.-Q. (1995). Acta Cryst. C51, 1412–1414.

Yang, Q.-C., Huang, D.-M., Chen, H.-Y. & Tang, Y.-Q. (1996). Acta Cryst. C52, 470–472.

Acta Cryst. (1997). C53, 925-927

trans-Methyl 2,9-Bis(phenylsulfonyl)-1,2,3,4-tetrahydro-4-β-carbolinylideneacetate†

E. Subramanian,* L. Govindasamy and K. Ravichandran

Department of Crystallography and Biophysics, University of Madras, Guindy Campus, Madras 600 025, India. E-mail: crystal@giasmd01.vsnl.net.in

(Received 28 October 1996; accepted 4 February 1997)

Abstract

The title compound, $C_{26}H_{22}N_2O_6S_2$, consists of a β carboline moiety, two phenylsulfonyl groups and one methoxycarbonylmethylene group. The six-membered heterocyclic ring adopts a half-chair conformation. The substituent at the indole N atom is equatorial while that at the second N atom is axial. The methoxycarbonylmethylene group is approximately coplanar to the β carboline moiety. The dihedral angle between the two phenyl rings is 78.3 (1)°.

Comment

Derivatives of β -carbolines have been used as inverse agonist pharmacophores for the benzodiazepine receptor. While 3-substituted β -carbolines have been shown to interact at benzodiazepine receptors, a number of other related β -carbolines have proven to be effective non-competitive inhibitors of the indolamine 2,3-dioxygenase (IDO) enzyme system (Cox & Cook, 1995). The tetrahydro- β -carboline unit is central to an enormous range of indole alkaloids, and stereospecific syntheses of derivatives are widely sought (Everett *et al.*, 1990).

[†] DCB contribution No. 877.

Fig. 1 shows the displacement ellipsoid plot of the title molecule, (I), with the numbering scheme. The geometry at the S atoms (S1 and S2) is distorted from the ideal tetrahedral configuration. The C-N distances in the C ring [N2-C9 = 1.474(5), N2-C9 = 1.474(5), N2-C10 = 1.463(5)Å] are significantly higher than in the B ring [N1-C1 = 1.419(5), N1-C11 = 1.404 Å] in agreement with the electron-withdrawing character of the phenylsulfonyl group. The S-N bond distances [S1-N1 = 1.670(3), S2-N2 = 1.634(4) Å] lie within the observed range of 1.63-1.69 Å (Kálmán, Czugler & Argay, 1981). The mean S-O, S-C and S-N distances of 1.430(3), 1.759(4) and 1.652(4)Å, respectively, conform to those found in N-(4-nitrophenyl)-4-phenylsulfonamide (Gomes et al., 1993). The dimensions of the three phenyl rings (A, D and E) are normal. The phenylsulfonyl group bonded to N1 is in an equatorial position, while the substituent at N2 is in an axial position.





The rings A, B, D and E are planar, while A and B are also coplanar. The six-membered heterocyclic ring C is in a half-chair conformation. The dihedral angle between rings D and E is $78.3(1)^{\circ}$. A large number of C—H···O-type interactions appear to contribute to crystal packing and stability.

Experimental

The compound was synthesized by a reductive condensation process (Mohanakrishnan, 1995) in the Department of Organic Chemistry, University of Madras, India. The crystals were grown from acetone at room temperature.

Crystal data

 $C_{26}H_{22}N_2O_6S_2$ Mo $K\alpha$ radiation $M_r = 522.59$ $\lambda = 0.71069 \text{ Å}$ Triclinic Cell parameters from 25 ΡĪ reflections a = 10.956(1) Å $\theta = 15 - 25^{\circ}$ $\mu = 0.265 \text{ mm}^{-1}$ b = 14.626(1) Å c = 7.606(1) Å T = 293 (2) K $\alpha = 93.66(1)^{\circ}$ Transparent parallelepiped $\beta = 92.06(1)^{\circ}$ $0.25 \times 0.2 \times 0.15$ mm $\gamma = 92.59(1)^{\circ}$ Pale yellow $V = 1214.2(2) \text{ Å}^3$ Z = 2 $D_x = 1.429 \text{ Mg m}^{-3}$ D_m not measured Data collection Rigaku AFC-7S diffractom- $\theta_{\rm max} = 25.00^{\circ}$ $h = 0 \rightarrow 13$ eter $\omega/2\theta$ scans $k = -17 \rightarrow 17$ $l = -9 \rightarrow 8$ Absorption correction: none 4260 measured reflections 3 standard reflections 4058 independent reflections every 100 reflections 2771 reflections with frequency: 60 min $I > 2\sigma(I)$ intensity decay: <1% $R_{\rm int} = 0.0177$ Refinement Refinement on F^2 $(\Delta/\sigma)_{\rm max} = 0.024$ $\Delta \rho_{\rm max} = 0.366 \ {\rm e} \ {\rm \AA}^{-3}$ $R[F^2 > 2\sigma(F^2)] = 0.044$ $wR(F^2) = 0.166$ $\Delta \rho_{\rm min} = -0.416 \ {\rm e} \ {\rm \AA}^{-3}$ S = 1.074Extinction correction: 4058 reflections SHELXL93 (Sheldrick, 404 parameters 1993) Extinction coefficient: All H-atom parameters refined 0.0039 (12) $w = 1/[\sigma^2(F_o^2) + (0.0509P)^2]$ Scattering factors from + 2.1063P] International Tables for

Table 1. Selected geometric parameters (Å, °)

Crystallography (Vol. C)

where $P = (F_o^2 + 2F_c^2)/3$

S

s

S

02	1.426 (3)	NI-C11	1.404 (5)
01	1.430 (3)	NI-CI	1.419 (5)
N1	1.670 (3)	N2-C10	1.463 (5)

Fig. 1. 30% probability displacement ellipsoid plot of the molecule with the numbering scheme.

S1-C21	1.758 (4)	N2	1.474 (5)
S2—O4	1.433 (3)	O5-C13	1.193 (6)
S2—O3	1.432 (3)	O6-C13	1.346 (6)
S2—N2	1.634 (4)	O6-C14	1.444 (6)
S2-C15	1.761 (4)		
C11N1C1	107.9 (3)	C2-C1-N1	129.6 (4)
C11—N1—S1	125.2 (3)	N2-C9-C8	113.8 (3)
C1-N1-S1	126.2 (3)	N2-C10-C11	108.5 (3)
C10-N2-C9	113.7 (3)	C7-C11-N1	109.5 (3)
C10-N2-S2	119.5 (3)	N1-C11-C10	126.3 (3)
C9N2S2	117.4 (3)	O5-C13-O6	123.3 (4)
C13-06-C14	116.0 (4)	O5-C13-C12	128.0 (4)
C6C1N1	107.7 (4)	O6C13C12	108.7 (4)
\$1—N1—C1—C2	-6.1 (7)	S2-N2-C9-C8	-90.3 (4)

The H atoms of C14 were fixed by a riding model. All the other H atoms were refined isotropically.

Data collection: Rigaku AFC-7R diffractometer. Cell refinement: MSC/AFC Diffraction Control Software (Molecular Structure Corporation, 1988). Data reduction: TEXSAN PRO-CESS (Molecular Structure Corporation, 1993). Program(s) used to solve structure: SHELXS86 (Sheldrick, 1985, 1990). Program(s) used to refine structure: SHELXL93 (Sheldrick, 1993). Molecular graphics: ORTEP92 (Vickovic, 1994). Software used to prepare material for publication: PARST (Nardelli, 1983, 1995).

The authors thank Professor J. Shashidhara Prasad (University of Mysore) for the use of the National datacollection facility.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: NA1278). Services for accessing these data are described at the back of the journal.

References

- Cox, E. D. & Cook, J. M. (1995). Chem. Rev. 95, 1797-1842.
- Everett, J. H., Reynolds, C. D., Sparks, C. A., Pangborn, W., Bailey, P. D., Dauter, Z., Helliwell, M. & Hollinshead, S. P. (1990). J. Crystallogr. Spectrosc. Res. 20, 109–115.
- Gomes, A. C., Biswas, G., Biswas, S., Biswas, G. K., Iitaka, Y. & Banerjee, A. (1993). J. Crystallogr. Spectrosc. Res. 23, 513-517.

Kálmán, A., Czugler, M. & Argay, G. (1981). Acta Cryst. B37, 868-877.

- Mohanakrishnan, A. K. (1995). PhD thesis, University of Madras, India.
- Molecular Structure Corporation (1988). MSC/AFC Diffractometer Control Software. MSC, 3200 Research Forest Drive, The Woodlands, TX 77381, USA.
- Molecular Structure Corporation (1993). TEXSAN. Single Crystal Structure Analysis Software. Version 1.6. MSC, 3200 Research Forest Drive, The Woodlands, TX 77381, USA.
- Nardelli, M. (1983). Comput. Chem. 7, 95-98.
- Nardelli, M. (1995). J. Appl. Cryst. 28, 659.
- Sheldrick, G. M. (1985). SHELXS86. Program for the Solution of Crystal Structures. University of Göttingen, Germany.
- Sheldrick, G. M. (1990). Acta Cryst. A46, 467-473.
- Sheldrick, G. M. (1993). SHELXL93. Program for Crystal Structure Refinement. University of Göttingen, Germany.
- Vickovic, I. (1994). J. Appl. Cryst. 27, 437.

Acta Cryst. (1997). C53, 927-928

N,N'-Dicyclohexylurea[†]

L. GOVINDASAMY AND E. SUBRAMANIAN*

Department of Crystallography and Biophysics, University of Madras, Guindy Campus, Madras 600 025, India. E-mail: crystal@giasmd01.vsnl.net.in

(Received 23 September 1996; accepted 26 February 1997)

Abstract

The title molecule, $C_{13}H_{24}N_2O$, possesses twofold symmetry, with the C=O group lying on a crystallographic twofold axis in the unit cell. The cyclohexane rings adopt a chair conformation. Successive molecules along the twofold axis are related by the short *b* translation and interact through two identical diad-related N-H···O hydrogen bonds [2.962 (5) Å] in which the O atom lying on the twofold axis acts as a common acceptor for the diad-related amino protons.

Comment

An interesting aspect of the title structure, (I), is that the asymmetric unit consists of mainly the N-cyclohexyl group and half of the C=O moiety. The C=O group lies on a crystallographic twofold axis and thus the title molecule possesses a twofold symmetry relating the N-cyclohexyl groups. The bond lengths and angles observed in the cyclohexane ring, as well as their variations around the ring, are found to be within the observed range of values reported for related structures, such as N,N'-dicyclohexyl-N-4-phenylbutyrylurea (Ball, Brown & Bennet, 1990) and 3-cyclohexyl-1-[2-(1*H*indol-3-yl)ethyl]urea (Ishida, Yamashita, Takai & Inoue, 1983).



An ORTEP diagram (Johnson, 1965; Vickovic, 1994) of the title molecule is shown in Fig. 1. For the cyclohexane ring, the average value of the bond angle, 112.8°, and that of the absolute torsion angle, 50.7° , as well as the distribution of the torsion angles around the ring, agree closely with the theoretically predicted values of 111.5 and 54.7° for chair conformations (Bixon & Lifson, 1967). As stated in the abstract, the molecules interact through two identical intermolecular N— $H \cdots O$ hydrogen bonds. This interaction gives rise to an infinite sheet-like structure, similar to the parallel

[†] DCB contribution No. 878.